

Predictions of Adsorption of Elements 112 and 114 on Various Surfaces

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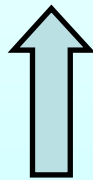
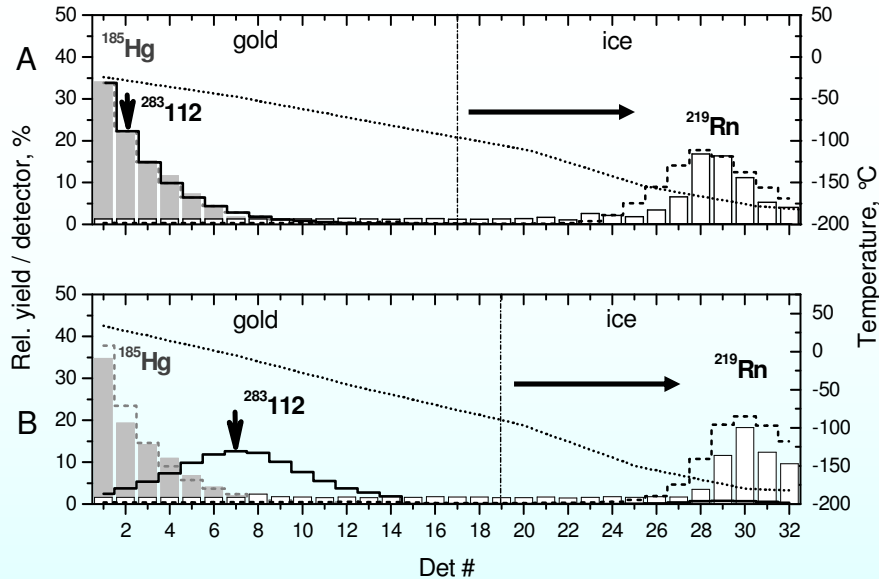
J. Anton and T. Jacob

University of Ulm

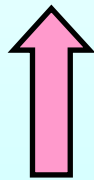
A. Borschevsky, E. Eliav and U. Kaldor

Tel Aviv University

Experimental Observations



112



114

$$T_{\text{ads}}(\text{exp.}) = -5\text{ }^{\circ}\text{C}$$

$$\Delta H_{\text{ads}}(\text{exp.}) = -52_{-4}^{+20}\text{ kJ/mol}$$

$$T_{\text{ads}} = -88\text{ }^{\circ}\text{C}$$

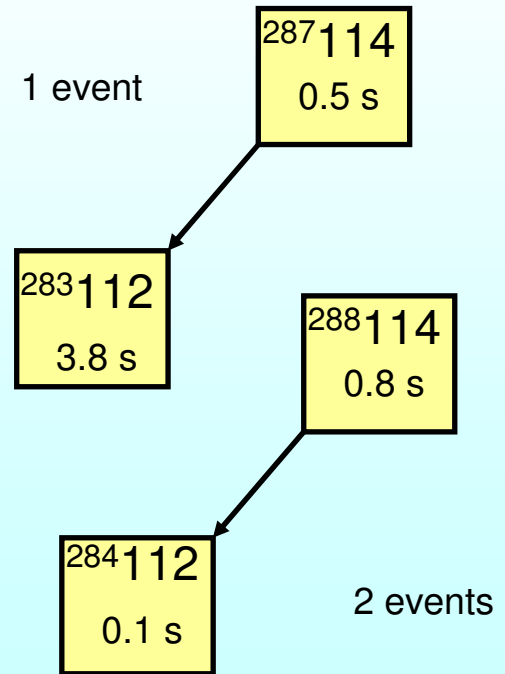
$$\Delta H_{\text{ads}} = -35_{-3}^{+19}\text{ kJ/mol}$$

112: R. Eichler, *et al.* Nature, **447**, 72 (2007)

114: Science News Online, April, 12, 2008;

Vol. 173, Nr. 15

R. Eichler, *et al.* NRC7, Budapest, 2008, Abstract



Atomic Properties of Elements 112 and 114 and Adsorption on Inert Surfaces

$$E(x) = -\frac{3}{16} \left(\frac{\epsilon - 1}{\epsilon + 2} \right) \frac{\alpha_{mol}}{\left(\frac{1}{IP_{slab}} + \frac{1}{IP_{at}} \right) R_{vdW}^3}$$

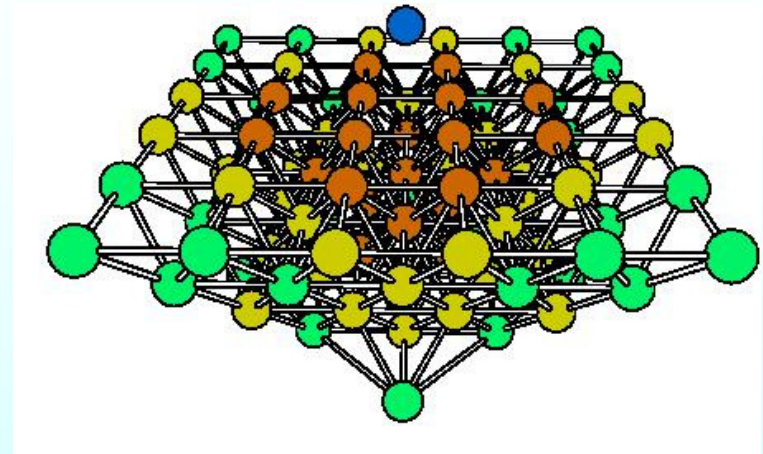
DCB CCSD(T) calculations (A. Borschevsky)

Property	112	114
Electronic configuration	d ¹⁰ s ²	s ² p _{1/2} ²
IP, eV	11.97	8.54
α, a.u.	27.4	29.5
AR, a.u.	3.21	3.30
R _{vdW} , a.u.	3.75	3.94
ΔH _{ads} (quartz), kJ/mol	27	21
ΔH _{ads} (ice), kJ/mol	26.2	20.2
ΔH _{ads} (Teflon), kJ/mol	16.4	10.4

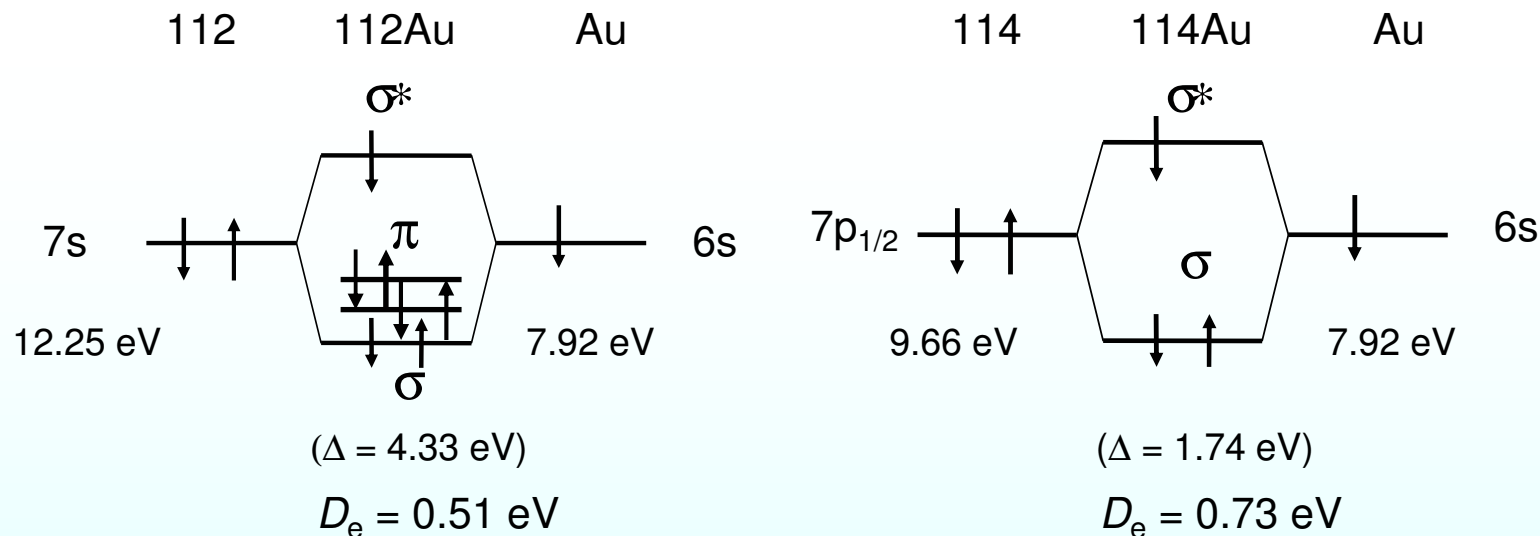
Predictions of Interaction of Elements Hg/112 and Pb/114 with Metals

4c - DFT calculations for:

- dimers MAu (V.P.)
- medium-size and large clusters MAu_n (J. Anton)
 - n=16 ... n=120
- embedded clusters (MAu_n)Au_m
 - n=34-36 m=156



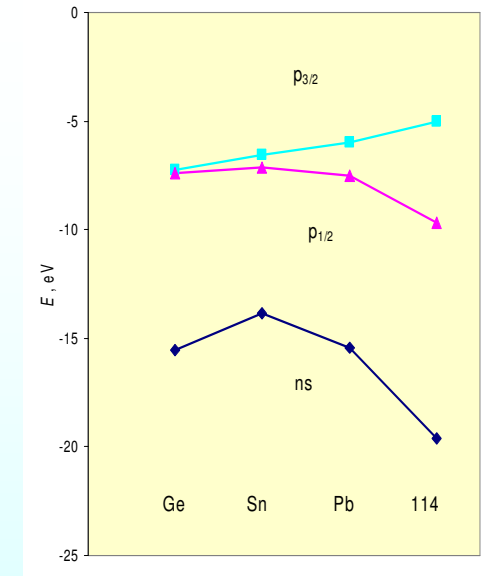
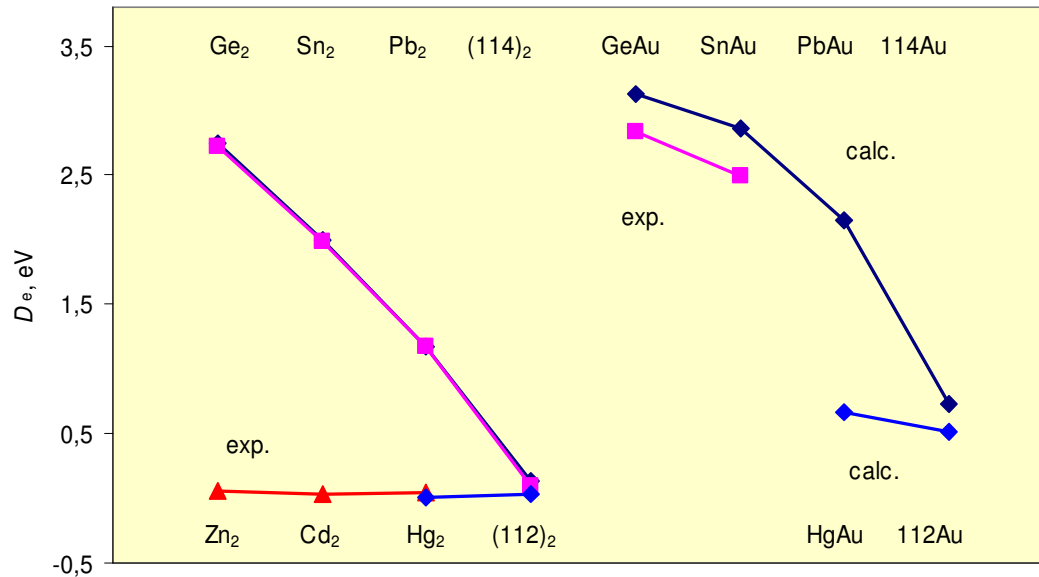
MO Energies and Composition of ^{112}Au and ^{114}Au



112Au			114Au		
MO	Energy, eV	Composition, %	MO	Energy, eV	Composition, %
σ_{112}	-3.007	(68) $7p_{1/2}$ (112)+(11) $6s$ (Au)+(8) $6p_{3/2}$ (Au)	π_{114}	-2.34	(94) $7p_{3/2}$ (114)+(2) $5d_{5/2}$ (Au)+(3) $6p_{3/2}$ (Au)
σ_{Au}^*	-5.885	(11) $7s$ (112)+(72) $6s$ (Au)+(9) $5d_{5/2}$ (Au)	σ_{114}^*	-4.935	(39) $7p_{1/2}$ (114)+(41) $6s$ (Au)+(9) $5d_{5/2}$ (Au)
π_{Au}	-6.542	(2) $6d_{5/2}$ (112) + (98) $5d_{5/2}$ (Au)	σ_{Au}	-5.797	(9) $7p_{1/2}$ (114)+(30) $6s$ (Au)+(57) $5d_{5/2}$ (Au)
π_{Au}	-6.651	(16) $6d_{5/2}$ (112)+ (84) $5d_{5/2}$ (Au)	π_{Au}	-5.880	(100) $5d_{5/2}$ (Au)
σ_{Au}	-6.756	(1.2) $7s$ (112)+(4) $6d_{5/2}$ (112) + (87) $5d_{5/2}$ (Au)+(4) $6s$ (Au)	π_{Au}	-6.123	(0.7) $7p_{3/2}$ (114)+(98) $5d_{5/2}$ (Au)+(0.5) $5d_{3/2}$ (Au)
Ground $^2\Sigma^+$		$d_{\text{Au}}^{10} \sigma_{\text{Au}}^2 \sigma_{\text{Au}}^* 1$	Ground $^2\Sigma$		$d_{\text{Au}}^{10} \sigma_{\text{Au}}^2 \sigma_{114}^* 1$

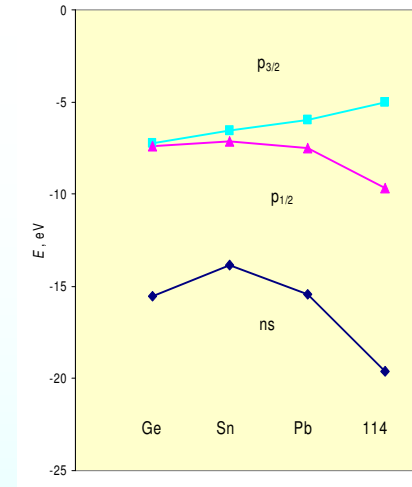
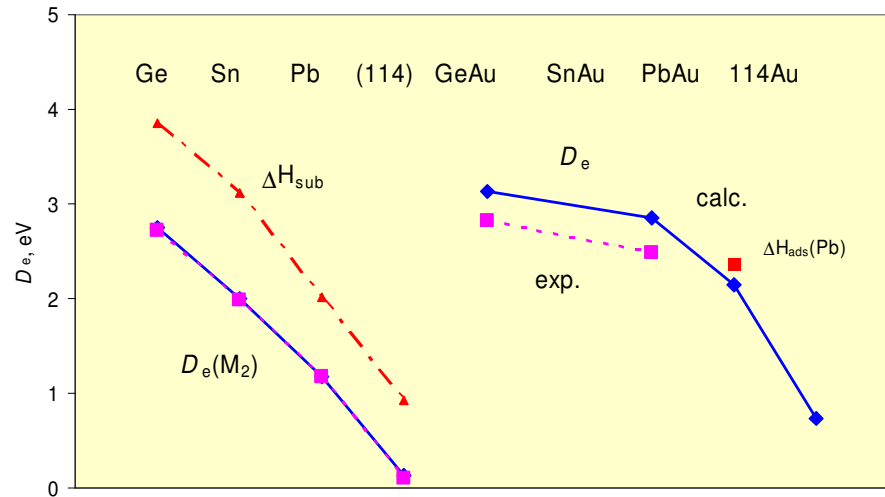
* HOMO

Comparison of Group 12 and 14 Dimers



Element 114 should be more reactive than 112. Large difference between Pb and element 114.

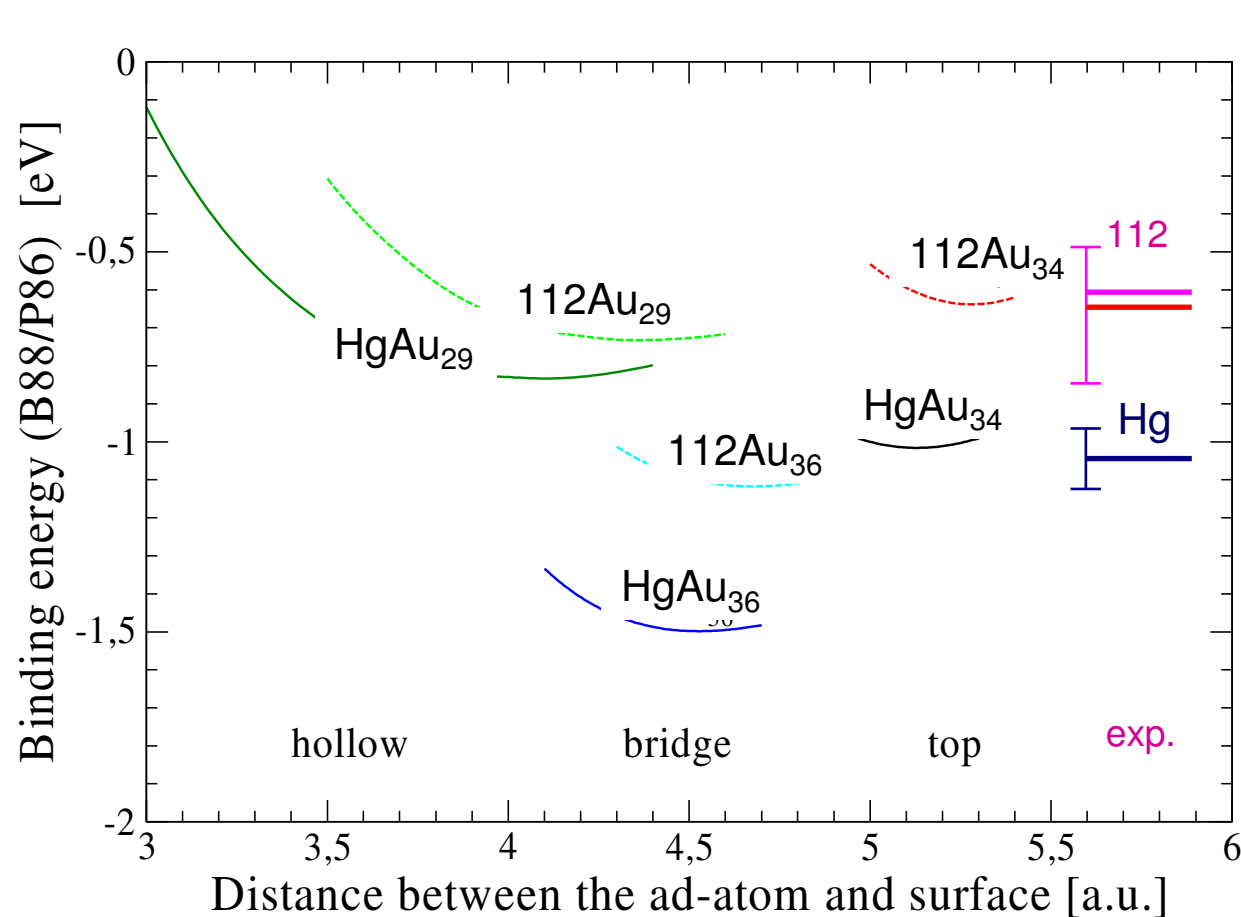
Correlation between $D_e(M_2)$ and $\Delta H_{\text{sub}}/\Delta H_{\text{ads}}$



Molecule	D_e , eV	ΔH_{ads} , eV	Molecule	D_e , eV	ΔH_{ads} , eV
Ge ₂	2.70	1.76	GeAu	3.14	-
Sn ₂	2.00	1.18	SnAu	2.86	-
Pb ₂	1.17	2.02	PbAu	2.15	2.37
(114) ₂	0.13	0.93	114Au	0.73	(0.95)
		$0.74 \pm 0.16^*$			$(0.97)^*$

* H. Rossbach and B. Eichler

Results of Embedded Cluster Calculations



112

Predicted:

$T_{\text{ads}}(\text{calc.}) = 0 \text{ } ^\circ\text{C}$

$\Delta H_{\text{ads}}(\text{calc.}) = -62 \text{ kJ/mol}$

[V. Pershina *et al.*,
Nucl. Phys. A **734**, 200 (2004);
ibid **787**, 381 (2007)]

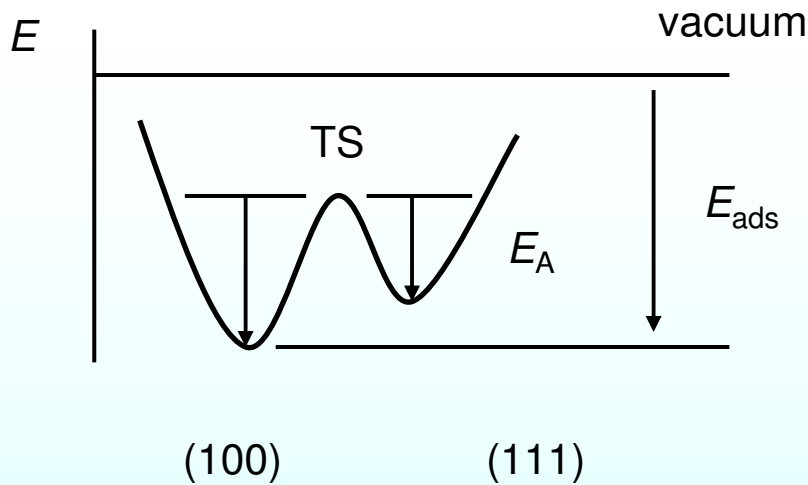
Observed:

$T_{\text{ads}}(\text{exp.}) = -5 \text{ } ^\circ\text{C}$

$\Delta H_{\text{ads}}(\text{exp.}) = -52_{-4}^{+20} \text{ kJ/mol}$

[R. Eichler, *et al.*
Nature, **447**, 72 (2007)]

Comparison of Au(100) and Au(111) Surfaces

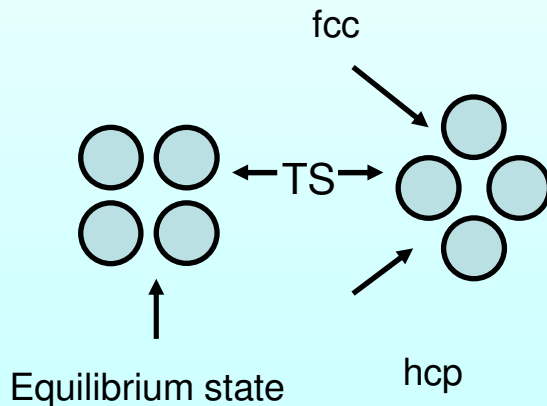


Surface reconstruction

Au(100) face-centered-cubic (fcc)



hexagonal-close-packed (hcp)



Surface	E_A , eV
Au(100)	0.62
Au(111)	0.22
Δ	0.40